

## New Anthracycline Antibiotics 10-*epi*-Oxaunomycin and 10-*epi*-11-Deoxyoxaunomycin

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Two new anthracycline antibiotics, 10-*epi*-oxaunomycin and 10-*epi*-11-deoxyoxaunomycin, were photochemically obtained from anthracycline metabolites D788-1 (10-carboxy-13-deoxocarminomycin) and D788-3 (10-carboxy-11-deoxy-13-deoxocarminomycin) and were examined for their growth inhibitory activities on cultured L1210 leukemic cells. Effects of the S configuration of C-10 and a hydroxyl group at C-11 on the bioactivity are discussed in comparison with oxaunomycin and 11-deoxyoxaunomycin.

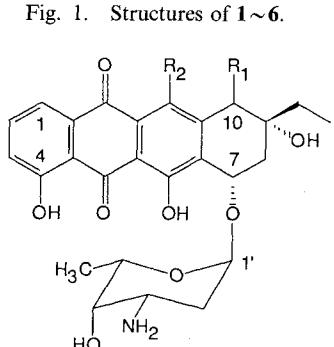
In the course of a screening program to obtain new anthracycline antitumor antibiotics, we have discovered oxaunomycin (1) (Fig. 1), a potent antibiotic of the rhodomycin-group as a minor component in the culture broth of a daunorubicin-blocked mutant<sup>1)</sup>. Afterward, we found that it was effectively formed by photochemical conversion of the anthracycline metabolite D788-1 (2)<sup>2)</sup>, which is an abundantly accumulated precursor in the culture of a daunorubicin-blocked mutant<sup>3)</sup>. 1 showed

very strong cytotoxicity against murine leukemia L1210 cells and a significant antitumor effect in some experimental tumors<sup>1)</sup>. However, it was too toxic for further development. Derivatization of **1** is one way to modify the toxicity without loss of its good antitumor effects, and several derivatives have been synthesized<sup>4~8)</sup>. De-substitution and configuration change in **1** itself also seem to be another interesting approach.

In this paper, we describe the photochemical preparation of 11-deoxy- and 10-*epi*-analogs (**4**~**6**) from the precursor anthracyclines D788-1 (**2**) and D788-3 (**3**)<sup>9</sup> and their antitumor activities *in vitro*. 11-Deoxyoxaunomycin has already been disclosed in 1987 as a product of *Actinomadura roseoviolacea* Ru7062<sup>10</sup>), but no data were presented on their biological activities.

## Results and Discussion

D788-1 (**2**) is a precursor anthracycline which is produced by the daunorubicin-blocked mutant RPM-5 of *Streptomyces* sp. D788<sup>3)</sup>. It is a water soluble material with a carboxyl group at C-10, which is biosynthetically converted to 13-deoxocarminomycin by enzymatic decarboxylation. However, it has been found that photochemical decarboxylation of **2** is followed by hydroxylation at the same site. Thus, oxaunomycin (10-hydroxy-13-deoxocarminomycin) (**1**) occurs in the photochemical reaction. There is also the possibility that



	R <sub>1</sub>		R <sub>2</sub>
1	OH	(R)	OH
2	COOH	(R)	OH
3	COOH	(R)	H
4	OH	(S)	H
5	OH	(R)	H
6	OH	(S)	OH

the stereoisomeric product, 10-*epi*-oxaunomycin, is produced as a minor reaction product. Furthermore, it is certain that the photochemical treatment of D788-3 (11-deoxy D788-1) (**3**) provides 11-deoxyoxaunomycin and related isomer by analogy with **2**.

Starting materials **2** and **3** were obtained by culturing the daunorubicin-blocked mutants RPM-5 and KL-330, respectively. Their photochemical treatments were carried out in the acidic acetone solution containing iodine according to the method described previously<sup>2</sup>. The reaction products were assayed in detail by HPLC after extraction with  $\text{CHCl}_3$ , and their time-course productions were monitored as shown in Fig. 2. **2** afforded **1** with a final conversion yield of about 80%. We found that a concomitant conversion to 10-*epi*-oxaunomycin (**6**) also occurred in this reaction although its yield was as low as 15%. No other significant product was observed.

On the other hand, the photochemical conversion of **3** gave a major 10-*epi*-11-deoxyoxaunomycin (**4**) (25% yield) and a minor 11-deoxyoxaunomycin (**5**) (<2% yield). **1** and **2** have a stereochemical configuration of

7(*S*), 9(*R*) and 10(*R*)<sup>3,9</sup>. The production of **4** as the main reaction product is not due to the stereochemical configuration of **3**, since it was determined to be 7(*S*), 9(*R*) and 10(*R*). However, there are significant differences in the coupling constants of 7-H and 8-CH<sub>2</sub> between **2** and **3**, indicating that the conformation of ring A is distorted for **3**. We believe this is a reason why **3** preferentially gave the 10-epimer in the photochemical reaction. For the structural elucidation of **4~6**, the photochemical reaction was conducted on a large scale using 400 mg each of **2** and **3**. The products were purified by silica gel column chromatography, reverse-phase HPLC and silica gel TLC, and pure **4**, **5**, and **6** were obtained in yields of 32 mg, 3 mg, and 15 mg, respectively.

<sup>1</sup>H and <sup>13</sup>C NMR data for **4~6** are shown in Tables 1 and 2. Chemical-shift assignments were mainly carried out by <sup>1</sup>H-<sup>1</sup>H and <sup>1</sup>H-<sup>13</sup>C COSY, and HMBC. The chemical shifts of C-10 for **4~6** were quite different from those of their parent compounds ( $\delta$  58.68 for **3**<sup>9</sup>) and  $\delta$  53.2 for **2**<sup>3</sup>) and the molecular ion peaks of the products appeared at 28 mass units lower than the corresponding

Fig. 2. Photochemical reaction of **2** (A) and **3** (B).

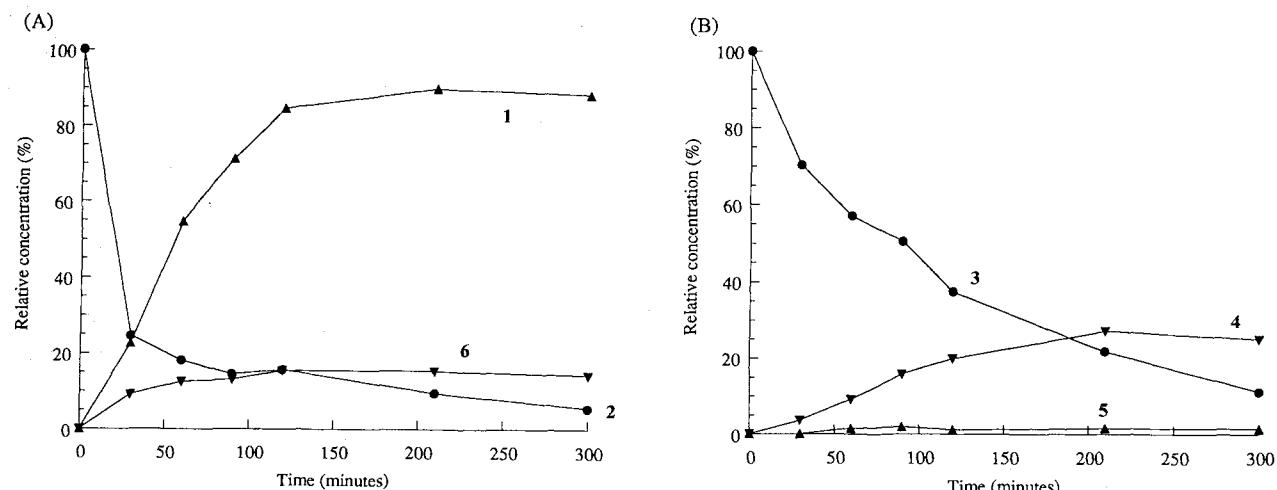


Table 1. <sup>1</sup>H NMR data for **4~6** and **1**.

Proton	<b>4</b>	<b>5</b>	<b>6</b>	<b>1<sup>a</sup></b>
7-H	5.11 br t (4.4, 2.9)	4.93	4.93 t (5.9)	5.15 dd (4.0, 2.0)
8-Ha	2.33 dd (14.7, 5.1)	2.27 dd (14.7, 5.1)	2.36 dd (13.9, 5.9)	2.23 dd (15.0, 2.0)
8-Hb	2.18 dd (14.7, 2.9)	2.18 d (13.9)	2.08 dd (13.9, 5.9)	2.13 dd (15.0, 4.0)
10-H	4.52 s	4.94 s	4.74 s	4.87 s
11-H	7.84 s	7.81 s	-	-
Solvent	$\text{CDCl}_3\text{-CD}_3\text{OD}$ (10:1)	$\text{CDCl}_3\text{-CD}_3\text{OD}$ (20:1)	$\text{CDCl}_3\text{-CD}_3\text{OD}$ (10:1)	$\text{CDCl}_3\text{-CD}_3\text{OD}$ (10:1)

<sup>a</sup> Data cited from ref 1.

Table 2.  $^{13}\text{C}$  NMR data for **4~6** and **1**.

Carbon	4	5	6	1
6a	133.54 <sup>a</sup>	134.08 <sup>a</sup>	136.89	134.77
7	71.01	73.47	71.57	70.75
8	33.50	33.71	35.69	32.48
9	73.08	72.33	72.19	71.94
10	72.82	66.76	67.91	66.09
10a	147.73	142.99	138.89	138.56
11	121.76	120.48	157.61 <sup>a</sup>	156.98 <sup>a</sup>
11a	130.65	132.06	112.08	111.49 <sup>a</sup>

<sup>a</sup> May be interchanged with similar one of other position (see experimental part). Solvent used was the same as shown in Table 1.

ones of the parent compounds, indicating that the carboxyl group is replaced by a hydroxyl group. The chemical shifts of 10-H for **4** and **6** shifted to upper fields when compared to **5** and **1**, respectively. The relative stereochemistry of **4~6** was examined by NOE experiments. NOEs were observed between 10-H and 13-CH<sub>2</sub> for **4**, **6**, and the aglycone of **4** (10-*epi*-11-deoxy- $\beta$ -rhodomycinone, **7**), although it was not observed for **5** and **1**. The coupling constants,  $J_{8-\text{Hb},7-\text{H}}$  for **4** and  $J_{8-\text{Ha},7-\text{H}}$  for **6** were larger than those of **5** and **1**, respectively. These observations suggest that **4~6** are 10-*epi*-11-deoxyoxaunomycin, 11-deoxyoxaunomycin, and 10-*epi*-oxaunomycin, respectively.

To elucidate the absolute stereochemistry, the configuration at C-9 of **3** was confirmed as follows. Treatment of **3** with CH<sub>2</sub>N<sub>2</sub>/ether gave a methylated compound (**8**) which was identified as 11-deoxy-13-deoxy-10-methoxycarbonylcarminomycin by MS and NMR analyses. Acid hydrolysis of **8** in 0.1 N HCl at 90° for 30 minutes afforded a yellow aglycone (**9**).  $^1\text{H}$  and  $^{13}\text{C}$  NMR data and molecular mass of **9** were identical to those of aklavinone<sup>11</sup> which has the stereochemical configuration of 7(S), 9(R) and 10(R). Furthermore,  $[\alpha]_D^{25}$  values of **9** (+130°, CHCl<sub>3</sub>) was in good agreement with that of aklavinone (+125°, CHCl<sub>3</sub>). These data demonstrated the configuration of **3** is 7(S), 9(R) and 10(R). The stereochemistry of **1** had been established to be 7(S), 9(R) and 10(R)<sup>3</sup>. From all these findings, the chemical structures of **4~6** were determined as shown in Fig. 1.

The antitumor activity of **1** and **4~6** *in vitro* against cultured L1210 cells was examined and the results are shown in Table 3. The cytotoxicities (IC<sub>50</sub> values) of **5** and **6** were low when compared with that of **1**. Especially, the configuration of C-10 greatly affected the reduced cytotoxicity. **4** also showed decreased cytotoxicity when

Table 3. Inhibitory activity of **4~6**, **1** and doxorubicin on growth of culture L1210 leukemia cells.

Compound	IC <sub>50</sub> ( $\mu\text{g/ml}$ )
	Growth
10- <i>epi</i> -11-deoxyoxaunomycin ( <b>4</b> )	0.037
11-Deoxyoxaunomycin ( <b>5</b> )	0.009
10- <i>epi</i> -oxaunomycin ( <b>6</b> )	0.48
Oxaunomycin ( <b>1</b> )	0.006
Doxorubicin	0.05

Cultured L1210 leukemia cells ( $5 \times 10^4$  cells/ml) were exposed for 48 hours to the drugs and the viable cells were counted by coulter counter. IC<sub>50</sub> is expressed as a drug concentration required to inhibit by a 50% of control growth of cultured L1210 cells.

compared with **5**. These results suggest that the epimerization of the hydroxyl group at C-10 plays an important role for the cytotoxic properties of this family of compounds. PENCO *et al.* reported the effects of conformations of ring A on antitumor activities in mice<sup>12</sup>. They found that 10(S)-methoxydaunorubicin had no activity on P388 lymphocytic leukemia although 10(R)-methoxydaunorubicin exhibited activity. This fact is similar to our observations. The coupling constants between 7-H and 8-CH<sub>2</sub> for **1** and **6** were almost the same as those of 10(R)- and 10(S)-methoxydaunorubicinones, respectively. Alternatively, the coupling constants for **4** were different from those for **6**, indicating that the conformation of ring A for **4** is distorted. Interestingly, the comparison of IC<sub>50</sub> values between **4** and **6** showed that the hydroxyl group at C-11 has no effect toward the reduction of cytotoxicity; since it is known that the absence of C-11 hydroxyl group results in considerably decreased cytotoxicity against cultured tumor cells<sup>9,13,14</sup>, we believe that the distorted ring A in **4** counterbalances the absence of hydroxyl group at C-11. Since **4** showed a moderate activity, equivalent to adriamycin, we expect it to be an improved oxaunomycin derivative with high antitumor activity *in vivo*.

## Experimental

### General

MP's were determined on a Kofler hot stage microscope and are uncorrected. UV spectra were recorded on a Hitachi U-3210 spectrophotometer and IR spectra (KBr pellet) on a Jasco FT/IR-5300 spectrometer.  $^1\text{H}$  and  $^{13}\text{C}$  NMR were recorded with a JEOL JNM-GSX400 spectrometer at 400 MHz and 100 MHz, respectively. Chemical shifts were expressed in  $\delta$  values (ppm) using TMS as an internal standard and coupling constants

were given in  $J$  (Hz). Mass spectra were recorded with a Hitachi JEOL JMS-SX102A spectrometer. Specific rotations were recorded on a Jasco DIP-181 digital polarimeter. TLC analyses were performed on Silicagel 60  $F_{254}$  (E. Merck) using solvent systems of  $\text{CHCl}_3$  -  $\text{MeOH}$  -  $\text{H}_2\text{O}$  -  $\text{AcOH}$  (30:10:0.4:0.2, system A) and  $\text{CHCl}_3$  -  $\text{MeOH}$  - concd  $\text{NH}_4\text{OH}$  (40:10:0.1, system B) as developing solvents.

#### HPLC Analysis

The concentration or purity of the products was determined using a reversed-phase HPLC method, which was performed on a Shimadzu HPLC system consisting of LC-6A pump, SPD-6AV detector, and Chromatopak C-R3A integrator with a analytical column, YMC-A312 (ODS) (150 × 6 mm i.d., Yamamura Chemical Laboratories Co., Ltd.). Solvent system of 30%  $\text{CH}_3\text{CN}$  (adjusted to pH 2.0 with phosphoric acid) was used as a mobile phase at a flow rate of 1.0 ml/minute, and UV absorbance was monitored at 254 nm. Retention times of **1~6** were 7.09, 5.81, 5.17, 6.51, 5.83, 12.03 minutes, respectively.

#### Photochemical Reaction of **2** and **3**

**2** or **3** (5 mg) was dissolved in 10 ml of 80%  $\text{Me}_2\text{CO}$  in 0.1 M citrate buffer (pH 3.5) and 5 mg of iodide were added to it. After adjusting to pH 5.2, the solution was exposed to a mercury lamp (UVL-400H-300P, Riko Science Industry; distance: 8 cm) in a water bath (25~30°C). The reaction solution (0.5 ml) was diluted with 2 ml of acidic water (pH 2.0, phosphoric acid) and 10  $\mu\text{l}$  aliquote was injected for HPLC analysis.

#### Preparation of **4** and **5**

Using 400 mg of **3** as a starting material, the photochemical reaction was carried out for 5 hours as described above. To the reaction solution were added 400 ml of  $\text{H}_2\text{O}$ , adjusted to pH 8.0, and the products were extracted with 800 ml of  $\text{CHCl}_3$ . The  $\text{CHCl}_3$  layer was washed with 200 ml of saturated aqueous  $\text{NaCl}$ , concentrated *in vacuo* to a small volume, and precipitated with an excess of *n*-hexane to give 150 mg of crude powder containing **4** and **5**. This crude powder was applied to a column of silica gel (20 mm i.d., Wakogel C-200, 25 g) and eluted stepwise with  $\text{CHCl}_3$  -  $\text{MeOH}$  - concd  $\text{NH}_4\text{OH}$  (120:10:0.05, 80:10:0.1, 40:10:0.1). Each of crude fractions enriched in **4** and **5** was further purified by preparative HPLC (column: Capcell Pak C18 SG120, 5  $\mu\text{m}$ , 250 × 30 mm i.d., Shiseido Co, Ltd.; mobile phase: 30%  $\text{CH}_3\text{CN}$  (adjusted to pH 2.0 with phosphoric acid), flow rate: 5 ml/minute, detection: UV at 254 nm). The fraction containing pure product was washed with toluene and the product was extracted with  $\text{CHCl}_3$  at pH 8.0. The  $\text{CHCl}_3$  layer was dried over anhydrous sodium sulfate and concentrated *in vacuo* to a small volume. To the concentrate, an excess of *n*-hexane was added to precipitate a yellowish orange powder. This procedure yielded 32 mg of **4** and 3 mg of **5**. **4**: MP

185~190°C (dec);  $[\alpha]_D^{20} + 131^\circ$  (*c* 0.02,  $\text{CHCl}_3$ ); IR (KBr)  $\text{cm}^{-1}$  1671, 1622; UV  $\lambda_{\text{max}}^{90\% \text{MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 229 (738), 258 (443), 433 (222); FAB-MS  $m/z$  500 ( $\text{M} + \text{H}$ )<sup>+</sup>,  $\text{C}_{26}\text{H}_{29}\text{NO}_9$ ; <sup>1</sup>H NMR ( $\text{CDCl}_3$  -  $\text{CD}_3\text{OD}$ , 20:1)  $\delta$  7.84 (1H, s, 11-H), 7.80 (1H, d,  $J$  = 7.34 Hz, 1-H), 7.68 (1H, t,  $J$  = 7.34 Hz, 2-H), 7.29 (1H, d,  $J$  = 7.33 Hz, 3-H), 5.42 (1H, d,  $J$  = 3.67 Hz, 1'-H), 5.11 (1H, br t,  $J$  = 4.40 and 2.93 Hz, 7-H), 4.52 (1H, s, 10-H), 4.12 (1H, q,  $J$  = 6.60 Hz, 5'-H), 3.49 (1H, br s, 4'-H), 3.03 (1H, br d,  $J$  = 11.74 Hz, 3'-H), 2.23 (1H, dd,  $J$  = 14.67 and 5.13 Hz, 8-Ha), 2.18 (1H, dd,  $J$  = 14.68 and 2.93 Hz, 8-Hb), 1.7~1.83 (2H, 2'- $\text{CH}_2$ ), 1.7~1.8 (1H, 13-Ha), 1.63 (1H, m,  $J$  = 7.34 Hz, 13-Hb), 1.33 (3H, d,  $J$  = 5.86 Hz, 6'- $\text{CH}_3$ ), 1.06 (3H, t,  $J$  = 7.34 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR ( $\text{CDCl}_3$  -  $\text{CD}_3\text{OD}$ , 20:1)  $\delta$  192.65 (C-5), 181.75 (C-12), 162.43 (C-4), 161.52 (C-6), 147.73 (C-10a), 137.38 (C-2), 133.54\* (C-6a), 133.12\* (C-12a), 130.65 (C-11a), 124.85 (C-3), 121.76 (C-11), 120.23 (C-1), 115.87 (C-4a), 114.70 (C-5a), 101.11 (C-1'), 73.08 (C-9), 72.82 (C-10), 71.01 (C-7), 70.35 (C-4'), 67.45 (C-5'), 46.42 (C-3'), 33.50 (C-8), 32.86 (C-2'), 29.32 (C-13), 16.83 (C-6'), 6.55 (C-14), similar values asterisked may be interchanged; Rf value: 0.22 (system A), 0.15 (system B). **5**: MP 145~147°C (dec);  $[\alpha]_D^{20} + 70^\circ$  (*c* 0.02,  $\text{CHCl}_3$ ); IR (KBr)  $\text{cm}^{-1}$  1630, 1607; UV  $\lambda_{\text{max}}^{90\% \text{MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 230 (643), 257 (422), 435 (196); FAB-MS  $m/z$  500 ( $\text{M} + \text{H}$ )<sup>+</sup>,  $\text{C}_{26}\text{H}_{29}\text{NO}_9$ ; <sup>1</sup>H NMR ( $\text{CDCl}_3$  -  $\text{CD}_3\text{OD}$ , 20:1)  $\delta$  7.87 (1H, d,  $J$  = 6.60 Hz, 1-H), 7.81 (1H, s, 11-H), 7.71 (1H, t,  $J$  = 7.34 Hz, 2-H), 7.34 (1H, d,  $J$  = 8.07 Hz, 3-H), 5.29 (1H, d,  $J$  = 3.66 Hz, 1'-H), 4.94 (1H, s, 10-H), 4.93 (1H, 7-H), 4.07 (1H, q,  $J$  = 6.60 Hz, 5'-H), 3.49 (1H, br s, 4'-H), 3.09 (1H, br d,  $J$  = 12.47 Hz, 3'-H), 2.27 (1H, dd,  $J$  = 14.67 and 5.14 Hz, 8-Ha), 2.18 (1H, d,  $J$  = 13.94 Hz, 8-Hb), 1.87 (1H, m,  $J$  = 7.34 Hz, 13-Ha), 1.84 (1H, t d,  $J$  = 13.94 and 3.67 Hz, 2'-Ha), 1.72 (1H, m,  $J$  = 7.34 Hz, 13-Hb), 1.67 (1H, dd,  $J$  = 13.94 and 4.40 Hz, 2'-Hb), 1.35 (3H, d,  $J$  = 6.60, 6'- $\text{CH}_3$ ), 1.11 (3H, t,  $J$  = 7.34 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR ( $\text{CDCl}_3$  -  $\text{CD}_3\text{OD}$ , 20:1)  $\delta$  187.86 (C-5), 187.57 (C-12), 162.66 (C-4), 162.18 (C-6), 142.99 (C-10a), 137.02 (C-2), 134.08\* (C-6a), 133.21\* (C-12a), 132.06 (C-11a), 125.17 (C-3), 120.48 (C-11), 119.62 (C-1), 116.15 (C-4a), 115.42 (C-5a), 98.53 (C-1'), 73.47 (C-7), 72.33 (C-9), 70.25 (C-4'), 67.70 (C-5'), 66.76 (C-10), 46.12 (C-3'), 33.71 (C-8), 33.20 (C-2'), 30.25 (C-13), 16.88 (C-6'), 6.56 (C-14), similar values asterisked may be interchanged; Rf value: 0.24 (system A), 0.20 (system B).

#### Hydrolysis of **4**

**4** (15 mg) was dissolved in 5 ml of 0.1 N HCl and hydrolyzed at 90°C for 30 minutes. The resulting precipitate consisting mainly of the aglycone was purified by preparative TLC (Silicagel 60  $F_{254}$ , E. Merck) using a developing solvent of  $\text{CHCl}_3$  -  $\text{MeOH}$  (15:1). The yellow band ( $R_f$  = 0.38) was scraped from the plates and eluted with  $\text{CHCl}_3$  -  $\text{MeOH}$  (10:1). After washing with  $\text{H}_2\text{O}$ , the eluate was concentrated *in vacuo* to a small volume and precipitated with an excess of *n*-hexane to give 11-deoxy-10-*epi*- $\beta$ -rhodomycinone (**7**) as a yellow powder (7 mg). **7**: MP 189~193°C (dec);  $[\alpha]_D^{20} + 86^\circ$  (*c*

0.02,  $\text{CHCl}_3$ ); IR (KBr)  $\text{cm}^{-1}$  1672, 1622, 1574; UV  $\lambda_{\text{max}}^{90\% \text{ MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 228 (966), 258 (593), 432 (300); FAB-MS  $m/z$  371 ((M + H)<sup>+</sup>,  $\text{C}_{20}\text{H}_{18}\text{O}_7$ ), 353 (M + H -  $\text{H}_2\text{O}$ )<sup>+</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub> - CD<sub>3</sub>OD, 10:1)  $\delta$  7.81 (1H, dd,  $J$  = 7.33 and 1.10 Hz, 1-H), 7.87 (1H, s, 11-H), 7.71 (1H, dd,  $J$  = 8.43 and 7.69 Hz, 2-H), 7.31 (1H, dd,  $J$  = 8.43 and 1.10 Hz, 3-H), 5.20 (1H, dd,  $J$  = 5.13 and 3.66 Hz, 7-H), 4.51 (1H, s, 10-H), 2.25 (1H, dd,  $J$  = 14.66 and 5.12, 8-Ha), 2.09 (1H, ddd,  $J$  = 14.66, 3.66 and 0.73 Hz, 8-Hb), 1.78 (1H, m, 13-Ha), 1.61 (1H, m,  $J$  = 7.33 Hz, 13-Hb), 1.07 (3H, t,  $J$  = 7.32 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub> - CD<sub>3</sub>OD, 2:1)  $\delta$  192.77 (C-5), 181.82 (C-11), 162.43 (C-4), 160.82 (C-6), 147.45 (C-10a), 137.46 (C-2), 133.62\* (C-6a), 132.83\* (C-12a), 124.85 (C-3), 121.72 (C-11), 120.29 (C-1), 115.89 (C-4a), 114.72 (C-5a), 73.50 (C-9), 72.94 (C-10), 62.81 (C-7), 34.56 (C-8), 28.96 (C-13), 6.48 (C-14), similar values asterisked may be interchanged.

### Methylation of 3

To a solution of 3 (50 mg) in 15 ml of MeOH were added 5 ml of CH<sub>2</sub>N<sub>2</sub>/ether and stirred at 25°C for 30 minutes. After dilution with 10 ml of 1% NaHCO<sub>3</sub>, the methylated product was extracted with 20 ml of CHCl<sub>3</sub> and concentrated *in vacuo* to dryness. The product was purified by preparative TLC using a developing solvent of CHCl<sub>3</sub> - MeOH - H<sub>2</sub>O - AcOH - concd NH<sub>4</sub>OH (120:50:5:1:1). The yellow band was scraped from the plates and eluted with CHCl<sub>3</sub> - MeOH (5:1). To the eluate was added H<sub>2</sub>O, washed with CHCl<sub>3</sub> and extracted with CHCl<sub>3</sub> at pH 8.0. The organic layer was concentrated *in vacuo* to a small volume and precipitated with an excess of *n*-hexane to afford 15-*O*-methyl-D788-3 (8) as a yellow powder (39 mg). 8: FAB-MS  $m/z$  542 ((M + H)<sup>+</sup>,  $\text{C}_{28}\text{H}_{31}\text{NO}_{10}$ ); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  7.75 (1H, d,  $J$  = 7.33 Hz, 1-H), 7.63 (1H, s, 11-H), 7.63 (1H, t,  $J$  = 7.33 Hz, 2-H), 7.26 (1H, d, 3-H), 5.44 (1H, br s, 1'-H), 5.22 (1H, br s, 7-H), 4.11 (1H, q,  $J$  = 7.34 Hz, 5'-H), 4.09 (1H, s, 10-H), 3.70 (3H, s, 16-CH<sub>3</sub>), 3.49 (1H, br s, 4'-H), 3.12 (1H, br d,  $J$  = 9.53 Hz, 3'-H), 2.51 (1H, dd,  $J$  = 15.41 and 4.41 Hz, 8-Ha), 2.32 (1H, d,  $J$  = 15.41 Hz, 8-Hb), 1.7~1.8 (3H, 13-Ha and 2'-CH<sub>2</sub>), 1.50 (1H, m,  $J$  = 7.33 Hz, 13-Hb), 1.33 (3H, d,  $J$  = 6.60 Hz, 6'-H), 1.08 (3H, t,  $J$  = 7.34 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  192.53 (C-5), 181.11 (C-12), 171.22 (C-15), 162.50 (C-4), 162.02 (C-6), 142.60 (C-10a), 137.26 (C-2), 133.38 (C-12a), 132.78 (C-6a), 131.19 (C-11a), 124.73 (C-3), 120.89 (C-11), 120.05 (C-1), 115.70 (C-4a), 114.56 (C-5a), 101.38 (C-1'), 71.61 (C-9), 71.00 (C-7), 70.31 (C-5'), 66.98 (C-4'), 57.05 (C-10), 52.48 (C-16), 46.33 (C-3'), 33.77 (C-8), 32.27 (C-2'), 32.07 (C-13), 16.83 (C-6'), 6.63 (C-14).

### Hydrolysis of 8

8 (30 mg) was dissolved in 10 ml of 0.1 N HCl and hydrolyzed at 90°C for 30 minutes. The aglycone was extracted with CHCl<sub>3</sub> and purified by preparative TLC using a developing solvent of CHCl<sub>3</sub> - MeOH (20:1). The yellow band was scraped from the plates and eluted with

CHCl<sub>3</sub> - MeOH (10:1). After washing with H<sub>2</sub>O, the eluate was concentrated *in vacuo* to a small volume and precipitated with an excess of *n*-hexane to give aklavinone (9) as a yellow powder (19 mg). 9: MP 197~200°C (dec);  $[\alpha]_D^{20} + 125^\circ$  (c 0.02, CHCl<sub>3</sub>); IR (KBr)  $\text{cm}^{-1}$  1734, 1674, 1624; UV  $\lambda_{\text{max}}^{90\% \text{ MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 229 (501), 258 (294), 289 (116), 431 (149); FAB-MS  $m/z$  413 ((M + H)<sup>+</sup>,  $\text{C}_{22}\text{H}_{20}\text{O}_8$ ); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  12.74 (1H, s), 11.97 (1H, s), 7.84 (1H, d,  $J$  = 7.61 Hz, 1-H), 7.73 (1H, s, 11-H), 7.71 (1H, t,  $J$  = 7.34 Hz, 2-H), 7.32 (1H, d,  $J$  = 7.33 Hz, 3-H), 5.39 (1H, br s, 7-H), 4.10 (1H, s, 10-H), 3.71 (3H, s, 16-CH<sub>3</sub>), 2.55 (1H, dd,  $J$  = 14.68 and 5.14 Hz, 8-Ha), 2.27 (1H, d,  $J$  = 14.67 Hz, 8-Hb), 1.73 (1H, m,  $J$  = 7.34 Hz, 13-Ha), 1.58 (1H, m,  $J$  = 7.33 Hz, 13-Hb), 1.10 (3H, t,  $J$  = 7.34 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  192.84 (C-5), 181.27 (C-12), 171.27 (C-15), 162.66 (C-4), 161.14 (C-6), 142.62 (C-10a), 137.52 (C-2), 133.55 (C-12a), 132.86 (C-6a), 132.70 (C-11a), 124.85 (C-3), 121.36 (C-11), 120.29 (C-1), 115.76 (C-4a), 114.74 (C-5a), 71.69 (C-9), 62.60 (C-7), 56.71 (C-10), 52.52 (C-16), 34.81 (C-8), 32.39 (C-13), 6.69 (C-14).

### Preparation of 6

The photochemical reaction of 2 (400 mg) was carried out in the same manner as described for 3, and yielded 220 mg of crude powder of 6. This crude powder was applied to a column of silica gel (20 mm i.d., Wakogel C-200, 25 g) and eluted stepwise with CHCl<sub>3</sub> - MeOH - H<sub>2</sub>O (100:10:0.5, 100:20:0.5) and CHCl<sub>3</sub> - MeOH - H<sub>2</sub>O - AcOH - concd NH<sub>4</sub>OH (120:50:5:1:1). 6 was further purified by preparative TLC using a developing solvent of CHCl<sub>3</sub> - MeOH - H<sub>2</sub>O - AcOH - concd NH<sub>4</sub>OH (120:50:5:1:1). The orange band was scraped from the plates and eluted with CHCl<sub>3</sub> - MeOH (5:1). To the eluate was added a half volume of 0.1 N acetate buffer (pH 3.0), and 6 was reextracted with CHCl<sub>3</sub> at pH 8.0. The organic layer was washed with H<sub>2</sub>O, dried over anhydrous sodium sulfate, and concentrated *in vacuo* to a small volume. To the concentrate, an excess of *n*-hexane was added to precipitate 6 as a red orange powder (15 mg). 6: MP 174~177°C (dec);  $[\alpha]_D^{20} + 17^\circ$  (c 0.02, CHCl<sub>3</sub>); IR (KBr)  $\text{cm}^{-1}$  1601; UV  $\lambda_{\text{max}}^{90\% \text{ MeOH}}$  nm ( $E_{1\text{cm}}^{1\%}$ ) 235 (969), 254 (515), 292 (185), 496 (338); FAB-MS  $m/z$  516 ((M + H)<sup>+</sup>,  $\text{C}_{26}\text{H}_{29}\text{NO}_{10}$ ); <sup>1</sup>H NMR (CDCl<sub>3</sub> - CD<sub>3</sub>OD, 10:1)  $\delta$  7.85 (1H, d,  $J$  = 8.07 Hz, 1-H), 7.71 (1H, t,  $J$  = 8.07 Hz, 2-H), 7.30 (1H, d,  $J$  = 8.80 Hz, 3-H), 5.42 (1H, br s, 1'-H), 4.93 (1H, t,  $J$  = 5.87 Hz, 7-H), 4.74 (1H, s, 10-H), 4.14 (1H, q,  $J$  = 6.60 Hz, 5'-H), 3.52 (1H, br s, 4'-H), 3.11~3.15 (1H, br m, 3'-H), 2.36 (1H, dd,  $J$  = 13.94 and 5.87 Hz, 8-Ha), 2.08 (1H, dd,  $J$  = 13.94 and 5.87 Hz, 8-Hb), 1.75~1.9 (2H, 2'-CH<sub>2</sub>), 1.62 (1H, m,  $J$  = 7.34 Hz, 13-Ha), 1.47 (1H, m,  $J$  = 7.34 Hz, 13-Hb), 1.31 (3H, d,  $J$  = 6.60 Hz, 6'-CH<sub>3</sub>), 1.01 (3H, t,  $J$  = 7.34 Hz, 14-CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub> - CD<sub>3</sub>OD, 10:1)  $\delta$  191.30 (C-5), 186.73 (C-12), 162.92 (C-4), 157.61\* (C-11), 157.18\* (C-6), 138.89 (C-10a), 137.66 (C-2), 136.89 (C-6a), 133.82 (C-12a), 125.32 (C-3), 120.18 (C-1), 116.45 (C-4a), 112.50 (C-5a), 112.08 (C-11a), 101.01 (C-1'), 72.19

(C-9), 71.57 (C-7), 70.33 (C-4'), 67.97 (C-5'), 67.91 (C-10), 46.97 (C-3'), 35.69 (C-8), 32.82 (C-2'), 30.51 (C-13), 17.24 (C-6'), 7.84 (C-14), similar values asterisked may be interchanged; Rf value: 0.18 (system A, in which 0.27 for 1), 0.07 (system B, in which 0.24 for 1).

### Biological Activity

Inhibitory effects of the drugs on growth in murine leukemia L1210 cell culture were determined according to the method as described previously<sup>15)</sup>.

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